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Through-conjugation of two phosphaalkyne ('C–P') moieties mediated by a bimetallic scaffold

Matthew C. Leech and Ian R. Crossley

Through-conjugation of two phosphaalkyne moieties within an isolable molecule is demonstrated for the first time with the synthesis of [(Ru(dppe)2)2(μ-(C≡C)2C6H4-p)(C≡P)2] via base-induced desilylation of [(Ru(dppe)2)2(μ-(C≡C)2C6H4-p)η1-C≡P=CSiMe3]2+. The nature of the cyaphide ligands and their influence upon the bimetallic core are studied electrochemically.

Phosphaalkynes (RC≡P) are archetypal models of the phosphorus–carbon analogy, being both isolobal and isoelectronic with alkynes. Though dichotomous in nature – by virtue of the polarity and lone-pair imparted by phosphorus – their chemical analogy to alkynes is well-established, with a prevalence of cycloaddition/oligomerisation reactions, while both η–alkynes) and η1-P (cf. nitriles, alkynyls) complexes with transition metals are known. Notwithstanding, an enduring omission lies with the incorporation of the discrete 'C≡P' moiety into architectures featuring extended conjugation (cf. the prevalence of polycyclitides), a desirable target – particularly from an organometallic standpoint – given extensive interest in acetylenic and phosphorus-containing moieties in the context of developing molecular electronic components.

Indeed, the conjugation of phosphaalkyne ('C≡P') moieties with other π-systems is limited to the small range of aromatic phosphaalkynes: PhC≡P, 2,6-R-C6H4C≡P (R = Mes, tBu), 2,6-R-4′-R′-C6H4C≡P (R = tBu, R′ = OMe, NMe2), R–R′ = tBu, CMe2Et and the putative P≡C–C≡E (E = CH, N, pTol, CO2Me, p-An). However, the latter (P≡C–C≡P) is also among a very limited range of compounds to feature two 'C≡P' moieties (Chart 1), and is the sole precedent example for which their mutual conjugation might reasonably be invoked (albeit unstudied).

Though a small number of transition metal complexes featuring trans-disposed η1-phosphaalkynes has been reported, viz. [M(L)2(P≡C′Bu)]2 (M = Mo, L = dppe, depe, R2PC6H4PR2, R = Tol, ClC6H4); [M(dppe)2(P≡CSiMe3)]2, even the concept of metal-mediated conjugation (cf. bis-alkynyl complexes) was explored prior to our recent report of the unprecedented cyaphide–alkynyl complexes trans-[Ru(dppe)2(p-C≡CR=C≡P)] (R = CO2Me, p-An). Herein, we extend this conceptual framework to consider, for the first time, extended conjugation between multiple 'C≡P' moieties, mediated by a bimetallic, redox-active, core; we also elucidate the electronic and redox nature of these complexes.

The sequential treatment of the bisethynylbenzene-bridged bimetallic complex [{Ru(dppe)2}2(μ-(C≡C)2C6H4-p)Cl2] (1) with two equivalents of AgOTf and P=CSiMe3 facilitates installation of two terminal phosphaalkyne moieties to afford 22+ (Scheme 1). Formation of 22+ is evident from characteristic spectroscopic signatures indicative of a coordinated phosphaalkyne (δP 111.4, δP 34 Hz) in proximity to the dppe scaffold (δP 42.2 (1 : 4 ratio)), while the carbon-rich bridge remains apparent from 13C{1H} NMR and infrared (νs=c 2054 cm−1) spectroscopic data. Retention of the silyl moieties follows from heteronuclear (1H–29Si) correlation, while the triflate counter-ion is observed in the 19F-NMR spectrum (δF ~78.9); bulk composition is confirmed by microanalysis.

The connectivity of 22+ is further supported by X-ray diffraction data (Fig. 1). The internal geometry is largely unremarkable, exhibiting only slight deviations from linearity about the metal centres (ζP–Ru–C 173.4(2), 175.3(2)) and in the bridge...
The optimized gas-phase geometries of $^{2+}$ and 3 (see ESI$^\dagger$) both exhibit slightly greater linearity about the metal centres and bridge when compared with the solid-state structure of $^{2+}$, alongside marginally longer $\rm C=\Pi$ linkages ($\sim$1.58 Å). These features are consistent with a prevalence of packing effects in the solid state, as noted previously for several $\eta^1$-$\Pi$-$\text{SiMe}_3$ complexes$^{20,23}$ and for our precedent cyphide–alkynyls.$^{16}$ The calculated $\Pi$-$\text{P}$ stretching mode for 3 (asym. $\nu_{\Pi\text{P}}$ = 1224 cm$^{-1}$) also compares well with experiment ($\nu_{\Pi\text{P}}$ = 1247 cm$^{-1}$). Notably, the experimentally observed frequency reflects a slightly stronger $\Pi$-$\text{P}$ linkage for 3 than in [RuH(dppe)$_2$(C=\Pi)P] ($\nu_{\Pi\text{P}}$ = 1239 cm$^{-1}$),$^{20\text{a}}$ attributable to competition with the trans-alkynyl for Ru $\rightarrow \pi^*$ donation. Indeed, we noted this previously for cyphide–alkynyls, though to a greater extent ($\nu_{\Pi\text{P}}$ = 1255, 1260 cm$^{-1}$),$^{16}$ suggesting a reduced competition within the bimetallic scaffold.

The frontier orbitals of $^{2+}$ and 3 (Fig. 2) show similarities, the HOMO in each case being dominated by the bridging $\pi$-system (76%, $^{2+}$; 54% 3) with a modest contribution from the metals (14% $^{2+}$; 26% 3). Notably, the HOMO of 3 also includes contributions from $\pi_{\text{C=\Pi}}$ (14%), which engage in out-of-phase mixing with the Ru ($d_{\Pi\text{P}-d_{\Pi\text{P}}}$, $\pi_{\Pi\text{P}-\Pi\text{C}}$, and $\pi_{\Pi\text{A}}$ orbitals, consistent with some level of through-conjugation. The contri-

### Table 1

<table>
<thead>
<tr>
<th></th>
<th>$\delta_{\Pi\text{P}}$</th>
<th>$\Delta \delta_{\Pi\text{P}}$</th>
<th>$\delta_{\text{C=\Pi}}$</th>
<th>$\Delta \delta_{\text{C=\Pi}}$</th>
</tr>
</thead>
<tbody>
<tr>
<td>$^{2+}$</td>
<td>111.4</td>
<td>198.9</td>
<td>281.8</td>
<td>92.0</td>
</tr>
<tr>
<td>3</td>
<td>159.7</td>
<td>281.8</td>
<td>281.8</td>
<td>93.7</td>
</tr>
<tr>
<td>[Ru(C$_2$R)(P=CSiMe$_3$)]$^+$</td>
<td>108.4</td>
<td>192.6</td>
<td>279.1</td>
<td>86.5</td>
</tr>
<tr>
<td>[Ru(C$_2$R)(C=\Pi)]P=CSiMe$_3$</td>
<td>112.8</td>
<td>188.2</td>
<td>281.9</td>
<td>93.7</td>
</tr>
<tr>
<td>[Ru(C$_2$R)(C=\Pi)]P=CSiPh$_3$</td>
<td>206</td>
<td>175.1</td>
<td>281.9</td>
<td>93.7</td>
</tr>
<tr>
<td>[RuH(P=CSiPh$_3$)]$^+$</td>
<td>165.0</td>
<td>21.3</td>
<td>281.9</td>
<td>93.7</td>
</tr>
<tr>
<td>$^{2+}$ [calc]$^\dagger$</td>
<td>118.4</td>
<td>188.8</td>
<td>279.1</td>
<td>86.5</td>
</tr>
<tr>
<td>3 [calc]$^\ddagger$</td>
<td>166.4</td>
<td>271.4</td>
<td>281.9</td>
<td>93.7</td>
</tr>
</tbody>
</table>

$^a$ [Ru] = Ru(dppe)$_2$. $^b$ $\Delta \delta$ on conversion from $\eta^1$-$\Pi$-$\text{SiMe}_3$ to cyphide.$^{16,20\text{a}}$ $^\dagger$ Increase in $\delta_{\Pi\text{P}}$ due to SiPh$_3$ vs. SiMe$_3$. $^\ddagger$ GIAO method with the PBE functional [anl1dz for Ru; 6-31G** for all other atoms]; referenced to H$_3$PO$_4$ or Me$_4$Si at the same level of theory.
butions from $\pi_{C=\pi}$ increase appreciably in the mutually degenerate HOMO-1 and HOMO-2 (~25%, see ESI†), lying 0.36 eV below the HOMO, albeit without involvement of the bridging arene (1%). In marked contrast, there is negligible contribution (<10%) from the $\eta^1-P\equiv CSiMe_3$ moieties of $2^+$ to any occupied frontier orbitals, their involvement becoming significant only in the appreciably stabilized HOMO-3 and HOMO-4, lying ca. 1.4 eV below the HOMO. Finally, in respect of $3$, we note that the terminal cyaphidic lone-pairs manifest in the HOMO-14 and HOMO-15, being stabilised by ca. 2 eV relative to the HOMO. This is entirely consistent with expectation, being similar to our previous observations,16 and those for phosphaalkynes more generally.24 Additionally, NBO calculations suggest these to reside in orbitals of ca. 75% s and 25% p character, as is typical of phosphaalkynes.

As is typical of complexes with the Ru(dppe)$_2$ scaffold, the latter dominates the virtual orbitals of $3$, which are mostly centred on the dppe ligands; the bridge contributes marginally to LUMO+12 and LUMO+14, lying 4 eV above the HOMO. In contrast, while the LUMO/LUMO+1 of $2^+$ are again dominated by the Ru(dppe)$_2$ framework, LUMO+2 is centred on the unsaturated core, with appreciable contributions from $\pi_{C=\pi}$ (60%) and the bridge (15%). This is reflected in the electronic spectrum of $2^+$, assigned in comparison with those derived from TD-DFT studies,25 calculating the first 200 excited states. This offers a fair approximation of the observed UV spectra for $2^+$ and $3$ (within limitations of the model), providing sufficient correlation to assist in the assignment of some key features. Thus, a feature at 350 nm (28 571 cm$^{-1}$) includes significant contribution from LLCT bands ($\pi_{C=C} \rightarrow \pi_{C=\pi}$ and $\pi_{C=C} \rightarrow \pi_{C=\pi}^*$) with marginal involvement of intraligand CT ($\pi_{C=C} \rightarrow \pi_{C=C}$), alongside the dominant MLCT and LLCT associated with excitation from the HOMO/HOMO-1 to low-lying dppe-based orbitals. A second feature around 260 nm (38 462 cm$^{-1}$) is primarily composed of ILCT within the dppe scaffold (~HOMO-10 $\rightarrow$ LUMO), but with additional contribution from $\pi_{C=\pi} \rightarrow \pi_{C=\pi}$ ILCT and $\pi_{C} \rightarrow \pi_{C=\pi}$ LLCT (HOMO-3 $\rightarrow$ LUMO+5).

In contrast, features in the UV/Vis spectrum of $3$ around 370 nm (27 027 cm$^{-1}$) and 250 nm (40 000 cm$^{-1}$) are wholly dominated by MLCT and LLCT transitions to the dppe scaffold, with marginal contributions from ILCT within the bridging $\pi$-framework; contributions from transitions to the high-lying $\pi^+_{C=\pi}$ (LUMO+36 to LUMO+39) are negligible.

The redox behaviours of $2^+$ and $3$ were explored using cyclic voltammetry (Table 2 and ESI†), both compounds exhibiting two distinct oxidative events, which can be assigned (trivially26) to sequential generation of the Ru$^{3+}$/Ru$^2+$ and Ru$^{3+}$/Ru$^{3+}$ species. For $2^+$ an initial quasi-reversible oxidation occurs at significantly more anodic potential than the corresponding (reversible) feature of $1$, presumably a corollary of its cationic nature. The second (irreversible) oxidation is similarly shifted to more positive potential,27 and demonstrates an appreciable stability for the mixed valence state $[2^{2+}]^+$, $K_c$ being comparable in magnitude to that of $[1]^{+}$ and related terminal alkynyls.19,28

In the case of $3$, two irreversible oxidations are observed, the initial event showing a slight anodic shift relative to $1$, and indeed related alkynyl systems;19,28 the second occurs at lower potential than the corresponding oxidation of $[1]^{+}$. On the reverse scan, an irreversible reduction process is observed at heavily cathodic potential. Notably, the diminished separation of the oxidative events indicates a reduced stability for the mixed valence state $[3]^{+}$ in comparison to $[1]^{+}$ and, indeed, related alkynyl complexes and $[2^{2+}]^+$, $K_c$ being two-orders of magnitude lower than for its counterparts.19,28 Notwithstanding, some stability is apparent, which implies some retention of the electronic coupling characteristic of the “Ru$_2$(μ–C≡C)$_2$C$_6$H$_4$P” scaffold, albeit diminished by the seemingly electron-acceptor character of the cyaphide ligand.

### Conclusions

In conclusion, we have described the first isolable compound to incorporate two ‘C≡P’ moieties as part of the same conjun-

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Table 2: Electrochemical (CV) data and comproportionation constants$^{ab}$

<table>
<thead>
<tr>
<th>$E_{pa}/V$</th>
<th>$E_{pc}/V$</th>
<th>$E_{1/2}/\Delta E_{pp}/V$</th>
<th>$\Delta E_{pa}/V$</th>
<th>$K_c$</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>−0.268</td>
<td>−0.348</td>
<td>−0.308 (80)</td>
<td>0.351</td>
</tr>
<tr>
<td>2$^{2+}$</td>
<td>0.081</td>
<td>0.004</td>
<td>0.043 (77)</td>
<td>0.290</td>
</tr>
<tr>
<td>3</td>
<td>0.995</td>
<td>0.565</td>
<td>0.635 (140)</td>
<td>0.190</td>
</tr>
</tbody>
</table>

$^a$ CH$_2$Cl$_2$/0.1 M [NBu$_4$]PF$_6$ using 1 mM analyte solutions at 25 °C, with Pt disc (1 mm) working electrode, Pt wire counter electrode and Ag wire pseudo-reference at 100 mV s$^{-1}$. Potentials relative to the Fe$^{3+}$/Fe$^{2+}$ couple (0.00 V), referenced using internal Fe$^{3+}$/Fe$^{2+}$ ($-0.56$ V ($E_{pp}$: 78 mV) vs. Fe$^{2+}$/Fe$^{3+}$). $^b$ $K_c = 10^{\Delta E/L} \text{mV}$ at 298 K. $^c$ Irreversible oxidation.

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Fig. 2: Frontier orbitals for $3$ (left) and $2^{2+}$ (right), with relative energies (see also ESI†).
gated scaffold, viz. [Ru₂{(μ{(C≡C)}}₂C₆H₄P)(C≡P)₂] (3). The electronic spectrum shows a dominance of LLCT and MLCT transitions from the bridge and phosphacarbon moieties to the dppe scaffold, with negligible ILCT within the π-system. The redox properties of 3 are more interesting and suggest some electron-acceptor character for the cyaphide ligand. While its electronic coupling within the bimetallic core provides initial conceptual validation for the incorporation of the cyaphide ligand into electro-active complexes. This will require engineering of appropriately stabilizing ancillary scaffolds, a challenge with which we are currently engaged.

**Conflicts of interest**

There are no conflicts to declare.

**Acknowledgements**

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**Notes and references**


22 Geometries were optimized from an initial model based on the solid state structure of $2^{2+}$, using the B3LYP functional (lanl2dz for ruthenium; 6-31G** for all other atoms). See ESI† for full details.


25 TD-DFT calculations were performed at the B3LYP level using lanl2dz for Ru and 3-21G* for all other atoms, without addition of a solvent model. Though a relatively low level of theory (constituting a balance against the complexity of the system and number of required excited states), such has previously proven suitable to provide a general guide to assignment.

26 Though commonly attributed to sequential Ru$^{II}$/Ru$^{III}$ couples, the oxidation events have heavy involvement from the carbon-rich bridge, due to extensive orbital mixing in the HOMO. These are thus more properly considered as sequential mono-oxidations of the bimetallic core.

27 Though mindful of previous reports of $1^+$ and related systems that describe the irreversible oxidation of $[1]^+$ close to $1^+$, the initial event being more consistent with a 1-electron process.