Transition metal catalyzed element–element additions to alkynes


This version is available from Sussex Research Online: http://sro.sussex.ac.uk/66514/

This document is made available in accordance with publisher policies and may differ from the published version or from the version of record. If you wish to cite this item you are advised to consult the publisher's version. Please see the URL above for details on accessing the published version.

Copyright and reuse:
Sussex Research Online is a digital repository of the research output of the University.

Copyright and all moral rights to the version of the paper presented here belong to the individual author(s) and/or other copyright owners. To the extent reasonable and practicable, the material made available in SRO has been checked for eligibility before being made available.

Copies of full text items generally can be reproduced, displayed or performed and given to third parties in any format or medium for personal research or study, educational, or not-for-profit purposes without prior permission or charge, provided that the authors, title and full bibliographic details are credited, a hyperlink and/or URL is given for the original metadata page and the content is not changed in any way.
Transition Metal Catalyzed Element-Element’ Additions to Alkynes

Melvyn B. Ansell*a, Oscar Navarro*b and John Spencer*a

aDepartment of Chemistry, University of Sussex, Brighton, BN1 9QI, U.K.
E-mail: ma462@sussex.ac.uk; j.spencer@sussex.ac.uk

bDivision of Biomaterials and Biomechanics, Department of Restorative Dentistry, School of Dentistry, Oregon Health & Science University, 2730 SW Moody Ave., Portland Oregon 97239, USA.
E-mail: navarrof@ohsu.edu

ABSTRACT
The efficient and stereoselective synthesis of, or precursors to multi-substituted alkenes has attracted substantial interest due to their existence in various industrially and biologically important compounds. One of the most atom economical routes to such alkenes is the transition metal catalyzed hetero element-element’π-insertion into alkynes.

This article provides a thorough up-to-date review on this area of chemistry, including discussions on the mechanism, range of E-E’ bonds accessible and the stoichiometric/catalytic transition metal mediators employed.

KEYWORDS: Alkynes, Alkenes, Element-Element bonds, Catalysis, Transition Metal

CONTENTS

1 Introduction ..................................................................................................................... 2
2 Transition Metal Mediated E-E’ Additions to Alkynes .................................................. 4
  2.1 Mechanism ............................................................................................................ 4
  2.2 Silicon-Silicon (Si-Si) ........................................................................................... 6
  2.3 Boron-Boron (B-B) ............................................................................................ 19
  2.4 Silicon-Boron (Si-B) ........................................................................................... 33
  2.5 Tin-Tin (Sn-Sn) ................................................................................................ 43
  2.6 Tin-Silicon (Sn-Si) ............................................................................................. 48
  2.7 Tin-Boron (Sn-B) ............................................................................................... 53
  2.8 Sulfur-Sulfur and Selenium-Selenium (S-S and Se-Se) ....................................... 55
  2.9 Sulfur-Silicon (S-Si) ........................................................................................... 62
  2.10 Sulfur-Boron (S-B) ............................................................................................ 62
1 Introduction

Alkenes are defined as either branched or unbranched hydrocarbons that possess at least one carbon-carbon double bond (C=C) and have a general formula of C\textsubscript{n}H\textsubscript{2n} [1]. Each carbon atom in a C=C bond is sp\textsuperscript{2}-hybridized, forming σ-bonds to three other atoms (Scheme 1). [2]. If the alkene has more than one substituent around the C=C bond then two geometric configurations are possible, E or Z (often termed trans or cis, respectively originating from the German words ‘entgegen’ meaning opposite and ‘zusammen’ meaning together (Figure 1).

![Figure 1](image)

**Figure 1** Alkene orbitals, shape and configurations for simple alkenes (olefins)

The importance of alkene stereochemistry is reflected in biologically relevant molecules and is often the difference between an active or inactive compound [3]. Furthermore, highly functionalized and stereodefined multi-substituted alkenes are found in many industrially important compounds including pharmaceuticals (Scheme 1.2) [4–8], dipeptide mimetics [9], and polymeric materials [10]. The stereoselective synthesis of, or precursors to, these alkenes has therefore attracted substantial attention from both academia and industry. A range of stereoselective syntheses are reported in the
literature, methods include Peterson olefination [11,12], the Ramberg-Bäcklund reaction [13], the Wittig reaction (as well as the Wittig-Horner variation) [14,15], olefin metathesis [16], Julia-Lythgoe olefination [17,18], and the McMurry reaction (Figure 2) [19,20].

**Figure 2** Selected examples of multi-substituted alkene synthesis and pharma-relevant alkenes

Arguably, one of the most atom economical routes (maximum number of atoms of the reactants appearing in the product/s) [21] in the synthesis of multi-substituted alkenes is alkyne reduction by its $\pi$-insertion into hetero element-element’ (E-E’, where E and E’
≠ H) bonds (albeit the newly formed C-E and C-E’ bonds are then further functionalized) [22]. These π-insertions result in the regio- and stereoselective formation of syn/cis-1-element-2-element’-alkenes in a single step (Scheme 1). A range of E-E’ bonds are accessible including Si-Si, Si-B, B-B, Sn-Sn, Sn-Si, Ge-Ge etc, and are mediated either stoichiometrically or catalytically by a variety of low-valent transition metal complexes.

\[
\begin{align*}
& R_1 = \text{alkyne} \\
& E-E' = \text{Si-Si, Si-B, B-B, Sn-Sn, Sn-Si, Ge-Ge...} \\
& \text{transition metal catalyst}
\end{align*}
\]

**Scheme 1** General scheme for transition metal mediated E-E’ additions to alkynes

Despite the chemical importance of this reaction and the industrial relevance of the 1-element-2-element’-alkenes as precursors to highly functionalized multi-substituted alkenes, the last thorough review on this topic was reported by Moberg and Beletskaya in 1999 (with a revised edition in 2006) [22,23]. A substantial volume of papers has since been published, and we believe there is a strong argument for an updated review that includes these more recent developments.

2 Transition Metal Mediated E-E’ Additions to Alkynes

2.1 Mechanism

The main mediators in the E-E’ additions to alkynes are low-valent platinum group transition metal complexes coordinated by either phosphine or isocyanide ligand sets. The mechanism is well established, both computationally and experimentally, and consists of three major steps: oxidative addition, insertion and reductive elimination.
The first step in this catalytic cycle is the oxidative cleavage of a E-E’ bond by a M(0)L₂ (M = platinum group metal, L = phosphine/isocyanide) species to form cis-(E)(E’)M(II)L₂ (MI1). MI1 is often kinetically stable and is isolated experimentally for many of the E-E’ bonds discussed above (the relevant E-E’ oxidative additions will be discussed in the appropriate chapters) [26–30]. A ligand exchange then occurs with decoordination of a single L ligand and coordination of the alkyne in its place to yield MI2. This is swiftly followed by an insertion of the alkyne into a M-E or M-E’ bond (MI3) [31,32]. The regioselectivity of the E-E’ addition is usually defined by this step and dictating factors include: the energetics of the bonds broken vs. the bonds formed, the sterics of the system and electronic stabilization effects within the resulting intermediates [33,34]. Experimental studies suggest that the insertion is often the rate-limiting step in these reaction pathways [31]. An isomerization and re-coordination of the L ligand results in the E and (E’) vinyl groups adjacent to one another. This positioning is then ideally suited for stereoselective reductive elimination to yield the corresponding Z-1,2-disubstituted alkenes and consequent reformation of M(0)L₂ (Scheme 2).
Scheme 2 General mechanism for platinum group transition metal mediated addition of E-E’ bonds to alkynes

This subsequent sections will overview the history, state-of-the-art and scope of this field, arranged by interelement bonds activated and metals used.

2.2 Silicon-Silicon (Si-Si)

Palladium

The $\pi$-insertion of unsaturated moieties into Si-Si bonds is often called bis(silylation). Palladium mediated bis(silylation) of alkynes is one of the most investigated reactions within this area of chemistry [35]. The first examples were reported by Kumada and Sakurai utilizing activated and strained disilanes, respectively. Kumada and co-workers demonstrated that activated disilanes, of the form $X_{3-m}Me_mSiSiMe_nX_{3-n}$ ($X = H, F, Cl$ or OMe; $m = 1-2$, $n = 1-2$), added to various alkynes when catalytic quantities of $[\text{Cl}_2\text{Pd} (\text{PR}_3)_2]$ or $[\text{Pd}(\text{PR'}_3)_4]$ ($R = \text{Et}$ or Ph and $R' = \text{Ph}$) were employed (Scheme 3) [36,37].
The extension of this protocol to non-activated disilanes, such as hexamethyldisilane (Me₃SiSiMe₃), was unsuccessful. Elsewhere, Sakurai showed that the extent of alkyne bis(silylation) using the strained cyclic disilane, 1,1,2,2-tetramethyl-1,2-disilacyclopentane, was dependent upon the choice of alkyne [38]. Dimethyl acetylenedicarboxylate, phenylacetylene and ethylene all underwent bis(silylation) (Scheme 3). However, no reaction was observed with the internal alkynes diphenylacetylene and bis(trimethylsilyl)acetylene.

\[
\text{Scheme 3} \text{ Kumada’s activated and Sakurai’s strained disilane bis(silylation)ations}
\]

Watanabe performed the bis(silylation) of acetylene using chlorinated disilanes, MeₙSi₂Cl₆₋ₙ (n = 2-5) [39]. The formation of the Z-1,2-disilylated alkenes was favored, although significant quantities of the E-isomers were noted. It was observed that upon heating, Z to E isomerization occurred in the presence of the Pd(0) complex. This work was extended to other activated disilanes such as methoxymethylidisilanes, (MeO)ₙMe₃₋ₙSiSiMe₃₋ₙ(OMe)ₙ, as well as the acetylenes 1-hexyne and
trimethylsilylacetylene [40]. Bis(silylation) with Me$_3$SiSiMe$_3$ was extremely sluggish even at temperatures of 140 °C.

Hiyama and co-workers utilized these chlorinated disilanes in the palladium catalyzed bis(silylation) of bis(trimethylsilyl)butadiyne [41]. Subsequent treatment of the reaction mixture with MeMgBr resulted in the formation of 1,1,4,4-tetakis(trimethylsilyl)butatriene (3) and/or 1,1,2,4-tetrakis(trimethylsilyl)-1-buten-3-yne (4) (Scheme 4).

Scheme 4 Bis(silylation) of diynes using chlorinated disilanes

The bis(silylation) of a number of internal and terminal alkynes using the activated disilane Me$_3$SiSiF$_2$Ph was achieved by Ozawa [42]. The catalyst was generated in situ from a the mixture of 1 mol% [Pd(η$^3$-allyl)Cl] and 2 mol% PMe$_2$Ph. Reactions were completed within several hours at room temperature giving the corresponding Z-alkenes. The choice of disilane was essential with no reactivity arising from the use of Me$_3$SiSiMe$_3$ or PhF$_2$SiSiF$_2$Ph.

Loy and co-workers employed the activated disilane, 1,2-dimethoxy-1,1,2,2-tetramethyldisilane in the bis(silylation) of 1,4-diethynylbenzene to form 5 (Scheme 5). 5 then ring closed at each alkenyl unit to form the corresponding disilaoxacyclopentenes. Subsequent hydrogenation with hydrogen gas using Pd on
carbon resulted in a saturated monomer that underwent ring-opening polymerization in
tetrahydrofuran (THF) or in the presence of catalytic quantities of tert-butylammonium
hydroxide (TBAH) giving rise to a crack free sol-gel in a matter of seconds [43].

Scheme 5 Bis(silylation) of 1,4-diethynylbenzene towards crack free sol-gels

Seyferth and co-workers demonstrated that the very reactive and strained Si-Si σ bond
in octamethyl-1,2-disilacyclobutane was capable of bis(silylating) a number of alkynes
including acetylene, phenylacetylene and dimethyl acetylenedicarboxylate when using
catalytic quantities of [Cl_2Pd(PPh_3)_2]. However, extension of this protocol to other
internal alkynes was unsuccessful, even at temperatures of 140 °C [44].

Manners showed that the ferrocenyldisilane, [Fe(η^5-C_5H_4)_(SiMe_3)_2] inserted into
acetylene or phenylacetylene to form the organometallic rings 6a and 6b, respectively
(Scheme 6) [45]. The reaction of alkynes such as dimethyl acetylenedicarboxylate
resulted in a mixture of mono- and di-insertion products with significant quantities of
the alkyne cyclotrimerization product, a common occurrence with alkynes such as
dimethyl acetylenedicarboxylate and acetylene [46]. Other palladium mediated
bis(silylations) of alkynes using strained disilanes have appeared in the literature: Ko’s
‘super-aromatic’ o-carborane disilane 7 [47], and Braunschweig’s
[2]silachromoarenophane 8 [48], bis(silylated) terminal and internal alkynes,
respectively (Figure 3). The cyclic nature of these disilanes pre-conditioned the formation of the Z-configured 1,2-disilylated alkene products.

Scheme 6 Palladium catalyzed ferrocenyl-disilane insertion into alkynes

Figure 3 Ko’s o-carborane disilane 7 and Braunschweig’s [2]silachromoarenophanes 8

In 1991, a communication from Ito and co-workers revolutionised the field of alkyne bis(silylation) by the introduction of the pre-catalytic combination of [Pd(OAc)$_2$] and isocyanide ligands [49]. As a result, the bis(silylation) of alkynes was no longer limited to activated or strained disilanes. A combination of 2 mol% [Pd(OAc)$_2$]/30 mol% tert-octyl isocyanide was enough to catalyze the bis(silylation) of terminal alkynes such as 1-phenylpropyne, 1-phenylhexyne, 1-nonyne and phenylacetylene using the non-activated disilane, Me$_3$SiSiMe$_3$. Reactions proceeded at 110 °C and resulted in unprecedented high stereoselectivities. Ito and co-workers extended this protocol to a range of bis(silylations) including the intramolecular bis(silylation) of alkynes in the stereoselective synthesis of 1,2,4-triols [50], cyclic tetrakis(organosilyl)ethenes as
organic chromophores [51], chiral allenylsilanes [52], and enantioenriched propargyl silanes [53].

Many authors since have utilized the \([\text{Pd(OAc)}_2]/\text{isocyanide}\) combination within their own work. For example, Strohmann and co-workers used the pre-catalytic combination above in a number of alkyne bis(silylations) using 1,1,2,2-tetramethyl-1,2-bis(phenylthiomethyl)disilane as the disilane source [54]. In particular, the bis(silylation) of ethynyl[2.2]paracyclophanes resulted in the formation of \(9\) (Scheme 7), which have potential applications in chiral catalysis and optoelectronic materials [55].

**Scheme 7** Bis(silylation) of ethynyl[2.2]paracyclophanes

The addition of \(\text{Me}_3\text{SiSiMe}_3\) to internal alkynes is considered one of the most challenging reactions in alkyne bis(silylation) chemistry. Even Ito’s \([\text{Pd(OAc)}_2]/\text{isocyanide}\) combination was catalytically inactive [49]. Navarro, Spencer and co-workers reported the synthesis of the novel compound \(\text{cis-[Pd(ITMe)_2(SiMe)_3]}\) (10, \(\text{ITMe} = 1,3,4,5\)-tetramethylimidazol-2-ylidene), via the oxidative addition of \(\text{Me}_3\text{SiSiMe}_3\) to \([\text{Pd(0)(ITMe)}_2]\) under mild conditions. 10 was subsequently employed as a pre-catalyst in the bis(silylation) of a range of 1,2-diarylalkynes using \(\text{Me}_3\text{SiSiMe}_3\) (Scheme 8) [56]. The resulting stilbenes 11 were synthesized with 100% Z-stereoselectivity and were either novel or previously only synthesized in a
stoichiometric manner. This unprecedented protocol was also extended to the bis(silylations) of two alkynes separated by a phenyl linker, to 1-silyl-2-aryl alkynes and to terminal alkynes.

Scheme 8 Bis(silylation) of internal alkynes with Me₃SiSiMe₃

Platinum

In contrast, the platinum catalyzed bis(silylation) of alkynes has been investigated to a lesser extent. The most common bis(silylation) mediator is [([η⁴-ethylene]Pt(PPh₃)₂]. Ishikawa detailed the bis(silylation) of a number of alkynes using 3,4-benzo-1,1,2,2,-tetra(isopropyl)-1,2-disilacyclobut-3-ene [57]. The reactivity and product selectivity using this platinum catalyst differed from the palladium analogues and depended on the alkyne used, notably employing extreme temperatures. Reactions with 1-hexyne and phenyacetylene resulted in a mixture of 12 and 13. The bulky mono-substituted alkynes
mesitylacetylene and (phenyldimethylsilyl)acetylene formed 13 as the sole product, whereas diphenylacetylene resulted in only the 1,2-disilylated alkene 14 (Scheme 9).

Scheme 9 Bis(silylation) of alkynes with 3,4-benzo-1,1,2,2,-tetra(isopropyl)-1,2-disilacyclobut-3-ene

Investigations into the reactivity of 1,2-bis(dimethylsilyl)carborane by Ko and co-workers were extended to the platinum catalyzed bis(silylation) of alkynes. Normal 1,2-bis(silylation) was observed in the reaction with phenylacetylene, diphenylacetylene, 3-hexyne, 2-butyne and dimethyl acetylenedicarboxylate. However, the use of 1-hexyne resulted in geminal or 1,1-bis(silylation) and the formation of a five-membered disilyl ring [58]. A later report by Ishikawa described the bis(silylation) of range of terminal and internal alkynes using cis- and trans-1,2-dimethyl-1,2-diphenyl-disilacyclopentane. The reactions proceeded with high stereospecificity and translation of the cis or trans nature of disilane in all cases [59].
The redox chemistry between gold(I)/(III) is similar to that of palladium(0)/(II), given that they are isolobal. This has triggered substantial research into the development of gold catalysts that are as active as their palladium analogues [60–62]. Despite this effort, gold catalysis is very much in its infancy with the only reports of alkyne bis(silylation) in the literature being mediated by gold nanoparticles supported on titanium oxide (Au/TiO₂) [63]. Stratakis and co-workers showed that the bis(silylation) of a range of terminal alkynes using hexamethyldisilane and 1,2-diphenyl-1,1,2,2-tetramethyldisilane was possible [64]. In all cases, the Z-alkenes were favored with a small percentage of the E-isomers formed. The heterogeneous catalyst gave comparable activities upon recycling. Stratakis extended the protocol to 1,1,2,2-tetramethyldisilane (HMe₂SiSiMe₂H). However, two isomers (15, major, and 16, minor) were isolated (Scheme 10). Mechanistically, this observation was explained by an initial bis(silylation) followed by a dehydrogenative addition to a second alkyne [65].

Scheme 10 Gold catalyzed bis(silylation)-dehydrogenative addition
Sunada and co-workers reacted 1,2-bis(dimethylsilyl)benzene with \([\text{Fe(mesityl)}_2]_2\) (mesityl = 2,4,6-\(\text{Me}_3\text{C}_6\text{H}_2\)) in aromatic solvents under a nitrogen (N\(_2\)) atmosphere to form 17 (Scheme 11). Subsequent addition of 2-butyne or phenylacetylene resulted in the quantitative formation of the disilacarbocycles 18\textit{a} and 18\textit{b}, respectively (Scheme 11) [66]. This process was made catalytic upon addition of 1,2-bis(dimethylsilyl)benzene to phenylacetylene and 20 mol\% of Fe [67]. Although this is not a bis(silylation) in the traditional sense (it lacks a Si-Si σ bond and it proceeds through a dehydrogenative double silylation), it is still a very rare example of an iron mediated bis(silyl)ation of alkynes.

**Scheme 11** Formation of a bis(silyl)Fe(II) complex and resulting reactivity with alkynes
Nickel

Kumada and co-workers accomplished the first examples of alkyne bis(silylation) using a nickel mediator. Bis(silyl)bipyridylnickel(II) complex 19 reacted with diphenylacetylene to form 20. Treatment of the latter with MeMgBr, followed by an acidic work up, resulted in the isolation of E-1,2-bis(trimethylsilyl)stilbene (21) (Scheme 12) [68]. Extension to other alkynes yielded mixtures of Z- and E-alkene products.

\[
\text{Scheme 12 Stoichiometric reaction of a bis(silyl)Ni(II) complex with diphenylacetylene}
\]

At the same time, Liu showed that a tetrafluorodisilacyclobutene underwent oxidative addition to [Ni(CO)\(_4\)]. The corresponding bis(silyl)Ni(II) complex 22 was reacted with tert-butylacetylene to form the two new 1,4-disilacyclohexadienes 23 and 24, where the \(^t\)Bu groups are syn and anti, respectively (Scheme 13) [69].
The first catalytic bis(silylation) of alkynes employing nickel was reported by Naka and co-workers. The reaction of 3,4-benzo-1,1,2,2-tetraethyl-1,2-disilacyclobutene with diphenylacetylene in the presence of catalytic amounts of [Ni(PEt$_3$)$_4$] formed the Z-alkene 25 [70]. As well as bis(silyl)ation, an alkyne insertion into one of the phenylene-Si bonds occurred, with 26 isolated as a minor product (Scheme 14). This type of insertion was consistently observed on applying the methodology to other alkynes [71].

**Scheme 13** Nickel mediated bis(silylation) using a strained cyclic disilane

![Scheme 13](image)

**Scheme 14** Ni(0) catalyzed bis(silylation) and phenylene-Si insertion

Rhodium

Examples of rhodium mediated alkyne bis(silylations) are rare. Tilley and co-workers carried out the stoichiometric reaction of [(Me$_3$P)$_3$RhSi(SiMe$_3$)$_3$] (27) with 2-butyne, resulting in the isolation of the Rh(III) complex 28 [72]. The authors proposed that 27 undergoes a facile silyl 1,2- and 1,3-migration (11 and 12, respectively) process in the presence of alkyne resulting in a [2+2] cycloaddition and the formation of a transient metallasilacyclobutene 13. The reductive elimination of a Si-C bond in 13 gives a Rh(I)-silyl intermediate 14, which then loses one PMe$_3$ ligand. This induces an oxidative addition of a Si-Si bond in the tethered trisilyl group and subsequent formation of 28 (Scheme 15).
Matsuda and co-workers studied the rhodium(I)-catalyzed intramolecular bis(silylations) of alkynes. Initial testing and optimization were executed on the disilanyl ether of a propargylic alcohol [73]. It was observed that 4-silyl-2,5-dihydro-1,2-oxasilole (29) was formed as the sole product (Scheme 16). This trans-bis(silylation) proceeded with the complete opposite stereoselectivity to the analogous palladium-catalyzed reaction. The protocol was then extended to a variety of (2-alkynylphenyl)disilanes affording the corresponding 3-silyl-1-benzosiloles (30) (Scheme 16).
2.3 Boron-Boron (B-B)

Platinum

Due to their low toxicity, high stability under atmospheric conditions and versatile reactivity, the synthesis of organoboron reagents has attracted significant interest. In particular, there is substantial focus towards Z-1,2-diborylated alkenes as the products of alkyne diboration [74]. The resulting newly formed B-C bonds are able to participate in Suzuki-Miyaura cross-coupling reactions [75], to build more complex and useful tri- and tetra-substituted alkenes. The first source of B-B bonds investigated was diboron tetrahalides. These contain the most reactive B-B bond available (the lack of π-donating substituents destabilizes the boron based p-orbital, this increases the Lewis acidity of the boron atoms and therefore their susceptibility towards nucleophilic attack) and often react with unsaturated organic substrates without the need for a transition metal mediator or catalyst [76]. However, the preparations of diboron tetrahalides are difficult.
and this therefore limits their synthetic utility [77,78]. Tetraalkoxy- and tetraaryloxydiborons are air stable, easily handled and, despite their relatively high B-B bond strengths, are now widely utilized in the stoichiometric and catalytic addition of B-B bonds to alkynes. These diboron reagents will be the main focus of this section.

Platinum is by far the most effective and widely studied mediator of alkyne diboration. Seminal results by Suzuki and Miyaura [79] indicated that 1-octyne inserted into the B-B bond of bis(pinacolato)diboron (B\(_2\)pin\(_2\)) using catalytic quantities of [Pt(PPh\(_3\))\(_4\)] to form 31 (Scheme 17).

\[ \text{Hex} \equiv \text{H} + \begin{aligned} \text{O} & \text{O} \\ \text{B} & \text{B} \\ \text{O} & \text{O} \end{aligned} \xrightarrow{3 \text{ mol\% [Pt(PPh\(_3\))\(_4\)]}} \begin{aligned} \text{O} & \text{O} \\ \text{B} & \text{B} \\ \text{O} & \text{O} \end{aligned} \text{DMF, 80 °C, 24 h} \quad 31 \]

**Scheme 17** The first diboration of alkynes catalyzed by [Pt(PPh\(_3\))\(_4\)]

The protocol was then extended to a range of internal and terminal alkynes with similarly high stereoselectivities obtained. The rate of diboration was drastically affected by the polarity of the solvent, with the more polar solvents (e.g. DMF) accelerating the rate. However, the authors later showed that hexane also accelerated the reaction rate and to a greater extent than most polar solvents [80]. Other transition metal complexes proved ineffective within this study, e.g. [Pd(PPh\(_3\))\(_4\)] and [Pd(OAc)\(_2\)]/isocyanide (the best catalysts in the bis(silylation) of alkynes). Suzuki-Miyaura’s coupling protocols have since been widely employed en route to, for example, enantiomerically enriched 1,2-diols [81], 5-benzylidenylbenzopyridyloxepine analogues as nuclear hormone receptors [82], 1H-phosphindoles as chiral helicenes.
10-mesitylborylsubstituted-dibenzoborepin as a photoresponsive material [84], and the pentacyclic alkaloid tylophorine [85].

Smith and co-workers carried out a stoichiometric diboration by reacting commercially available bis(catecolato)diboron (B$_2$cat$_2$) with [(η$^2$-4-octyne)Pt(PPh$_3$)$_2$]. This resulted in the oxidative addition bis(boryl)Pt(II) complex 32, and the Z-1,2-diborylated alkene 33 (Scheme 18) [86].

**Scheme 18** Platinum-mediated stoichiometric diboration of an alkyne

Marder and Norman extended the synthesis of bis(boryl)platinum(II) complexes to the use of other diborons including B$_2$pin$_2$ and B$_2$(4-t-Bucat)$_2$ (4-t-Bucat = 1,2-O$_2$-t-BuC$_6$H$_3$). [(η$^2$-ethylene)Pt(PPh$_3$)$_2$] and 34 were then used as catalysts in the diboration of terminal and internal alkynes employing B$_2$pin$_2$ and B$_2$cat$_2$ as B-B bond sources (Scheme 19).

**Scheme 19** Pt(0) and Pt(II) catalyzed diboration of internal and terminal alkynes
These catalysts were more efficient in the stereoselective formation of Z-1,2-diborylated alkenes than [Pt(PPh$_3$)$_4$], with reactions proceeding smoothly using 3 mol% of either catalyst at 80 °C. The rate and conversions were significantly affected by the choice of substituents on the alkyne and the diboron reagent. The presence of π-donating moieties on the alkyne resulted in faster reactions than π-withdrawing substituents and the fastest conversions proceeded in the order of B$_2$cat$_2$ > B$_2$pin$_2$ > B$_2$(4-Bu'cat)$_2$ [87].

The platinum catalyzed diboration of internal and terminal alkynes using diboron 1,2-B$_2$Cl$_2$(NMe$_2$)$_2$, affords cyclic 1-azonia-2-borata-5-boroles (36) (Scheme 20). The key feature within these structures was that the boron and nitrogen atoms exhibited both a three and four-coordinated center. Although the mechanism for forming 36 was unclear, the authors proposed an initial diboration followed by a rearrangement of the B-Cl and B-NMe$_2$ bonds [88].

![Scheme 20 Diboration-rearrangement of alkynes using 1,2-B$_2$Cl$_2$(NMe$_2$)$_2$](image)

Baker and co-workers developed a phosphine-free platinum catalyzed diboration of 1-octyne and di-p-methylphenylacetylene using B$_2$cat$_2$ [89]. The reactions proceeded using 5 mol% of the commercially available [Cl$_2$Pt(cod)] (cod = 1,5-cyclooctadiene) at 55 °C. This protocol was highly dependent on the choice of diboron source, with only B$_2$cat$_2$ accessible, as well as the choice of halide and diene on the platinum metal. [Br$_2$Pt(cod)] required pre-stirring for 24 h before a homogeneous catalytic mixture was
obtained and even then reaction yields were lower. The use of dicyclopentadiene instead of cod as a ligand also resulted in the formation of the 1,2-diborylated alkenes in lower yields.

In a study into new routes for the preparation of 1,1-geminal sp²-organobismetallic derivatives, Srebnik and co-workers demonstrated the platinum catalyzed diboration of 1-alkynylphosphonates and 1-alkynylboronates afforded the Z-1,2-diborylated vinylphosphonates and trisboronated alkene products 37 and 38, respectively (Scheme 21) [90]. The reaction with alkynylboronates was extremely sensitive to the moisture content of the solvent with ‘wet’ solvents resulting in B-C bond cleavage via a hydrodeboronation. Elsewhere, Nishihara reported the platinum catalyzed diboration of phenylethynyl MIDA (MIDA = N-methylimidiacetic acid) boronate with B₂pin₂ to form 1,1,2-triboryl-2-phenylethene [91].

![Scheme 21 Diboration of 1-alkynylphosphonates and 1-alkynylboronates](image)

Fernandez and co-workers reported the preparation of α,α’-difluorinated carbonyl compounds. The reactions proceeded by an initial platinum(0) catalyzed diboration of internal and terminal alkynes to form Z-1,2-diborylated alkenes. Subsequent work-up
with the electrophilic fluoro-deboronation agent 1-(chloromethyl)-4-fluoro-1,4-diazeniabicyclo[2.2.2]octane ditetrafluoroborate (or Selectfluor) resulted in formation of the $\alpha,\alpha'$-difluorinated carbonyl compounds 39 (Scheme 22) The stereochemistry of the diborylated alkene remained in the fluorinated carbonyls. If trace amounts of water were found within the solvent or reaction mixture the difluoromethyl alcohols 40 were isolated [92]. The authors later optimized this protocol to a one-pot diboration/fluoro-deboronation microwave procedure. This resulted in the shortening of reaction times to several minutes and the lowering of catalyst loadings to as little as 0.05 mol% [93].

Fernandez and co-workers extended their investigations into the use of N-heterocyclic carbene (NHC) [94–96] platinum complexes as catalysts. The platinum species 41 were formed by the transmetallation reaction between the corresponding NHC-silver compound and Karstedt’s catalyst (Scheme 23). Initial assessment of 41’s catalytic activity in the diboration of alkynes found that 41b, with triazoylidene carbene, was the
most active and suitable mediator for this reaction. A range of internal and terminal alkynes were then diborylated using $\text{B}_2\text{cat}_2$ and 5 mol% of $41b$ [97].

\[
\begin{align*}
\text{NHC-platinum-based catalysts in diboration of alkynes} \\
\text{Scheme 23} \\
\end{align*}
\]

Braunschweig and co-workers demonstrated that alkynes could insert into the B-B bond of [2]borametallarenophanes. These B-B bonds were deemed moderately strained, but thermally stable. The diboration was achieved stoichiometrically using $[\text{Pt(PPh}_3)_4]$ and 10 equivalents of 2-butyne to yield the ansa-bis(boryl)alkenes $42$ (Scheme 24) [98]. The diboration was also completed catalytically under both homogeneous and heterogeneous conditions over several days [99].

\[
\begin{align*}
\text{me}{\text{...}}^{\text{Me}} & \quad \text{[Pt(P}{\text{E}}_3)_3]} \quad \text{Me}{\text{...}}^{\text{Me}} \\
42 & \quad \text{5 mol% [Pt(P}{\text{E}}_3)_3]} \quad \text{or} \quad \text{6 mol% Pt sponge} \\
& \quad 80-100 ^\circ\text{C, 2-8 days} \\
& \quad n = 1 \text{ or } 2, M = \text{Fe or Cr}
\end{align*}
\]
The stereoselective, stepwise reactions of two non-equivalent boryl groups is highly desirable in the catalytic diboration of alkynes, as it may enable sequential couplings. Suginome and co-workers developed an unsymmetrical diboron, pinB-Bdan (pin = pinacolato; dan = naphthalene-1,8-diaminato) (Scheme 25) [100]. In the presence of phosphine-platinum catalysts, the diboration of terminal alkynes resulted in the regioselective formation of 43 with Bdan, a boryl protecting group, in the terminal position. The palladium-catalyzed Suzuki-Miyaura cross-coupling occurred chemoselectively on the more reactive internal Bpin. This was in sharp contrast to the B₂pin₂ based diborations, where the coupling selectively proceeds initially at the more reactive terminal Bpin group.

Escribano and co-workers showed that titania-supported platinum nanoparticles were efficient catalysts for the diboration of alkynes under solvent and ligand free conditions in air. Terminal and internal alkynes were accessible at 70 °C using 0.2 mol% of Pt/TiO₂. A range of electron-donating and withdrawing aromatic or alkyl, branched and
cycloalkyl substituents were accessible. Exclusively Z-1,2-diborylated alkenes were observed in all cases [101]. In contrast, when the support was magnesia (MgO), higher loadings and the use of solvent and elevated temperatures of 130 °C were required [102].

Palladium

Palladium-catalyzed diboration of alkynes are rare. Examples employing [2]borametalloarenophanes were reported by Braunschweig and co-workers. Palladium on carbon was utilized as a heterogeneous catalytic source, however the diborations required higher temperatures and much longer conversion times than their platinum analogues [98,99]. The rarity of palladium mediated alkyne diborations can be attributed to the energetics of the B-B bond oxidative addition at the Pd(0) center. Theoretical calculations suggest that this is both a kinetically and thermodynamically unfavorable process [103]. Despite this, Spencer, Navarro and co-workers recently accomplished the facile diboration of terminal and internal alkynes using catalytic amounts of [Pd(ITMe)$_2$(PhC≡CPh)] (44) [104]. This represented the first examples of alkyne diborations utilizing a homogeneous palladium catalyst. Both terminal and internal alkynes were accessible with 100% syn-stereoselective formation of the corresponding 1,2-diborylalkenes 45 (Scheme 26). These diborations proceeded with higher yields and/or under milder reaction conditions than their platinum analogues.
A computational study to determine the mechanistic route suggested that Pd(0) catalyzed alkyne diborations using NHC ligands followed the same catalytic cycle as phosphines (as detailed in the introduction of section 1.2) [104]. An integral process in this pathway was, as in the case of phosphines, the reversible dissociation of an NHC ligand. These NHC ligands were also key to a successful oxidative addition of the B-B bond at the Pd(0) center. They destabilize the (diboron)Pd(0)L₂ (L = ITMe) adduct resulting in a sufficient lowering of the free energy for oxidative addition.

Cobalt

In their investigations into the diboration of alkynes, Marder and co-workers described the diboration of 1,2-bis(4-(trifluoromethyl)phenyl)ethyne with B₂cat₂ (Scheme 27) using a [Co(PMe₃)₄] catalyst. Compound 46 was isolated as the major product of this reaction, with small quantities of the E-isomer detected [105].
Iron

The only example of iron catalyzed diboration of alkynes was detailed by Nakamura in 2015. Initial optimizations focused on the diboration of 4-octyne using B$_2$pin$_2$. Catalytic quantities of FeBr$_2$ and LiOMe with 1.5 equivalents of MeOBpin were enough to afford 47 in high yields [106]. On extending to other Fe(II) and Fe(III) catalysts, yields dramatically decreased. The diboration of a variety of internal alkynes was possible: those with alkyl substituents proceeded in high yields, whereas aryl or bulky alkyl groups retarded the diboration. The role of the additional borating agent was also assessed. In the absence of MeOBpin the reactions still proceeded, but with lower conversions. When using MeOBnep (MeOBnep = 2-methoxy-5,5-dimethyl-1,3,2-dioxaborinane) the unsymmetrical diborylalkene 48 was isolated as the major product (Scheme 28), suggesting that the incorporation of the second boryl unit was introduced by an electrophilic substitution reaction with MeOBnep or MeOBpin.
Scheme 28 Iron(II) catalyzed diboration of alkynes in a symmetric and unsymmetric manner

Iridium

Ozerov and co-workers devised a two-step reaction to convert alkynes into trisborylalkenes. The first step transformed terminal alkynes into alkynylboronates using pinacolborane (HBpin) and iridium complex 49 as a catalyst. Degassing this reaction mixture followed by the introduction of a CO atmosphere generated the new catalyst 50, which mediated the dehydrogenative diboration of the newly formed alkynylboronate with HBpin to form 51 (Scheme 29) [107]. This reaction was extended to a range of alkyl and aryl terminal alkynes. The authors proposed the reaction to proceed via a hydroboration intermediate or via $\text{B}_2\text{pin}_2$.
Scheme 29 Iridium catalyzed dehydrogenative borylation/diboration of terminal alkynes

Copper

Examples of group 11 transition metal catalyzed diboration of alkynes are very rare, with only one example of copper and one of gold described in the literature [108]. The first diboration of alkynes employing a copper catalyst was performed by Yoshida. The diboration of alkyl and aryl internal alkynes using B$_2$pin$_2$ in the presence of [Cu(OAc)$_2$] and PCy$_3$ resulted in high yields of the corresponding Z-1,2-diborylated alkenes [109]. The authors also extended this to the diboration of benzynes to form the resulting 1,2-diborylated benzenes (52) (Scheme 30). Changing the phosphine to P(Bu)$_3$, P(Oc)$_3$ or PPh$_3$ resulted in either prolonged reaction times or lower yields. A striking feature of this copper catalysis was the diboration of propargyl ethers. In all cases the tetraborolated product 53 was exclusively isolated (Scheme 30).
Gold

Nanoporous gold (AuNPore), prepared by dealloying the monolithic Au$_{30}$Ag$_{70}$ alloy in a 70% nitric acid electrolyte, is a highly active catalyst in the diboration of alkynes. Jin and co-workers optimized a system with phenylacetylene and B$_2$pin$_2$ utilizing 2 mol% of AuNPore at 100 °C [110]. The AuNPore catalyst was recyclable with no notable decrease in catalytic activity over multiple cycles. The protocol was extended to a variety of terminal and internal alkynes, however other diborons were ineffective. Mechanistically, the authors proposed absorption of the B$_2$pin$_2$ onto the AuNPore surface. The B-B bond is then cleaved at the low coordinate Au atoms to give an Au-Bpin species. The alkyne then adsorbs and reacts rapidly with two Au-Bpin species either through a simultaneous addition path to form the corresponding Z-adduct or in a stepwise manner.

Scheme 30 Copper-catalyzed diboration of benzyynes and tetraborylation of propargyl ethers
2.4 Silicon-Boron (Si-B)

Palladium

Silylboranes are attractive precursors in the element-element additions to unsaturated substrates such as alkynes. According to the Pauling scale, the electronegativity difference between the Si (2.12) and B (1.88) atoms [111], is such that 1-boryl-2-silyl alkenes are synthesised with chemo-, regio- and stereoselective control in a single transformation [112,113]. The boron and silicon functionalities in these alkene adducts can subsequently undergo chemoselective stepwise reactivity towards the preparation of more complex and unsymmetrical tri- and tetra substituted alkenes [114,115]. The most widely used catalysts for the silaboration of alkynes are group 10 transition metal complexes, specifically palladium-containing complexes.

Ito’s palladium/tert-alkyl isocyanide catalyst combination, previously detailed in alkyne bis(silylations), was effective in the silaboration of both terminal and internal alkynes to form syn-1-boryl-2-silyl alkenes (54) with high regio- and stereoselectivities (Scheme 31) [116]. The silylborane of choice was (dimethylphenylsilyl)boronic acid pinacol ester (PhMe₂SiBpin); this Si-B compound is thermally stable under inert conditions and the Bpin functionality improves the stability of the subsequent organo-compounds towards hydrolysis during purification. In the case of terminal alkynes, the silaboration proceeded with the addition of the boryl group at the terminal position. Attempts at silaboration employing other metal complexes resulted in either lower yields and mixtures of regioisomers (e.g. [Pt(PPh₃)₄]) or no activity (e.g. [RhCl(PPh₃)₃]).
The authors later extended this protocol to other silylboranes (i.e. PhMe₂SiB(NEt₂)₂ and PhMe₂SiBcat) and to a larger array of terminal and internal alkynes, including 1,7-octadiyne to afford the double silaboration product 55 (Scheme 31). The reactivity of the syn-1-boryl-2-silyl alkenes was also assessed. Suzuki-Miyaura cross-coupling and conjugate additions to methyl vinyl ketones at the alkenyl boryl group were possible, leading to 56 and 57, respectively (Scheme 32) [117]. Many authors have since utilized [Pd(OAc)₂]/isocyanide as a mediator in the silaboration of alkynes including in the synthesis of syn-homoallylic alcohols [118], multi-arylated olefins [119], and enamides [120].

**Scheme 31** Silaboration of alkynes employing a palladium/isocyanide catalyst
Tanaka and co-workers described the silaboration of 1-octyne employing the silylborane, 1,3-dimethyl-2-dimethylphenylsilyl-2-bora-1,3-diazacyclopentane. The corresponding Z-1-boryl-2-silyl alkene 58 was isolated by utilizing the pre-catalytic combination of \([\text{Pd}_2(\text{dba})_3]\) and epto (epto = 4-ethyl-2,6,7-trioxa-1-phosphabicyclo[2.2.2]octane, \(\text{P(OCH}_2\text{CEt})\)). Pre-heating the pre-catalytic combination at 80 °C for 5 minutes was necessary in order to generate the active catalytic species, proposed to be \([\text{Pd(epto)}_2]\). As observed in Ito’s report, the silaboration of terminal alkynes proceeded in a regioselective manner with the boryl group inserting at the terminal position. Low to no yields were observed on applying other phosphorus containing ligands such as \(\text{PMe}_3\) and \(\text{PPh}_3\). The protocol was also expanded to the silaboryl carbocyclization of hepta-1,6-diyne to form 59 (Scheme 33) [121].
Scheme 33 Silaboration and silaboryl carbocyclization of terminal alkynes

Pilot and co-workers synthesised stable organosilylboranes possessing mesityl groups on the boryl atom, (diphenylmethylsilyl)dimesitylborane \((\text{PhMe}_2\text{SiBMes}_2)\) and (diphenyl-tert-butylsilyl)dimesitylborane \((\text{Ph}_2\text{tBuSiBMes}_2)\). These silylboranes are not stabilized by electronegative groups on the boron atom e.g. oxygen or nitrogen, but instead through the steric bulk of the mesityl functionality. They were employed in the silaboration of terminal alkynes such as phenylacetylene, using the \([\text{Pd}_2(\text{dba})_3]/\text{ep}\text{to}\) catalytic combination. Steric clashing between the substituents of the alkyne and the boryl moiety precluded the silaboration of internal alkynes [122].

In their investigations into the silaboration of terminal alkynes, Suginome and co-workers showed that it was possible to tune the stereoselective preference of the reaction by altering the reagent stoichiometry. The reaction parameters were assessed on treating (chlorodimethylsilyl)pinacolborane \((\text{ClMe}_2\text{SiBpin})\) with 1-octyne in the presence of 1 mol% \([\eta^3\text{-C}_3\text{H}_5]\text{Pd(PPh}_3\text{Cl)}\), followed by subsequent addition of isopropyl alcohol (IPA) and pyridine. When excess 1-octyne was used the Z-isomer \(60\) was isolated as the sole product. However, excess \(\text{ClMe}_2\text{SiBpin}\) results in the formation
of the E-isomer 61 as the major product (Scheme 34). This observation was applicable to a range of terminal alkynes, although sterically hindered substituents on the alkyne restricted E-silaboration [123].

**Scheme 34**  Reagent dependent stereoselective silaboration of terminal alkynes

It was also possible to tune the regioselectivity in the silboration of terminal alkynes. The silaboration proceeds with ‘normal’ regioselectivity in the presence of catalytic quantities of \([(η^3-C_3H_5)Pd(CPPh_3)Cl]\). However, using the more sterically hindered phosphine \(\text{P(}^{Bu})_2\text{(biphenyl-2-yl)}\) an inverse or ‘abnormal’ regioselectivity was observed, with Z-2-boryl-1-silyl-1-alkenes 62 isolated as the major product (Scheme 35) [124].
Scheme 35 Ligand controlled, stereoselective ‘abnormal’ regioselective silaboration

\[
\begin{align*}
\text{Cl} & \quad \text{O} \\
\text{Si} & \quad \text{B} \\
\text{O} & \quad \text{O} \\
R_1 & \quad \text{H} \\
R_2 & \quad \text{R}'
\end{align*}
\]

\[
\text{MOH} \quad \text{toluene, 50 °C}
\]

\[
\begin{align*}
\text{Si} & \quad \text{O} \\
\text{B} & \quad \text{O} \\
R_1 & \quad \text{H} \\
R_2 & \quad \text{R}'
\end{align*}
\]

63a : \( R_1 = H, R_2 = n\text{Hex}, M = K \)
63b : \( R_1 = H, R_2 = n\text{Hex}, M = Na \)
63c : \( R_1 = H, R_2 = n\text{Hex}, M = Cs \)
63d : \( R_1 = H, R_2 = t\text{BuMe}_2\text{SiO(CH}_2\text{)}_3, M = K \)
63e : \( R_1 = H, R_2 = \text{Cl(CH}_2\text{)}_3, M = K \)
63f : \( R_1 = H, R_2 = \text{NC(CH}_2\text{)}_3, M = K \)
63g : \( R_1 = H, R_2 = \text{cyclohexyl}, M = K \)
63h : \( R_1 = n\text{Hex}, R_2 = H, M = K \)
63i : \( R_1 = t\text{BuMe}_2\text{SiO(CH}_2\text{)}_3, R_2 = H, M = Na \)
63j : \( R_1 = \text{Cl(CH}_2\text{)}_3, R_2 = H, M = Na \)
63k : \( R_1 = \text{NC(CH}_2\text{)}_3, R_2 = H, M = Na \)
63l : \( R_1 = \text{cyclohexyl}, R_2 = H, M = Na \)

\[
\begin{align*}
\text{K}^+ & \quad 1.5 \text{ eq. 4-idoanisole} \\
& \quad 5 \text{ mol\% Pd[P(tBu)_3]_2} \\
& \quad \text{H}_2\text{O} \\
& \quad \text{DMSO, 80 °C, 2h}
\end{align*}
\]

64a

\[
\begin{align*}
\text{pinBO-Si} & \quad \text{H} \\
\text{nHex} & \quad \text{H}
\end{align*}
\]

64

Scheme 36 Silylborate formation and resulting external-base free cross-coupling

Suginome and co-workers also hydrolysed the ‘normal’ and ‘abnormal’ silaborated alkenes with metal hydroxides MOH \((M = \text{Na or K})\) instead of the IPA/pyridine mixture. This resulted in the formation of a five-membered cyclic borate 63 via intramolecular attack of the resulting silanol oxygen with the tricoordinated boron atom.
The potassium borates 63a and 63h were then subjected to external-base free Suzuki-Miyaura cross-coupling with 4-iodoanisole to form 64 and 65, respectively (Scheme 36) [125].

The authors achieved a different mode of reactivity by substituting one of the substituents of the silicon atom of a silylborane for an amino group. The reaction of (Et₂N)Me₂SiBpin with aliphatic or aryl terminal alkynes resulted in the formation of 2,4- and 3,4-disubstituted siloles, 66 and 67 respectively (Scheme 37). Isomer 66 was favored in most cases and this was attributed to steric clashing within intermediates in the catalytic cycle. Deviations in the electronic and steric properties of the alkyne substituents had little influence on the regioisomer formed. However, altering the phosphine ligand to the more sterically hindered P(t-Bu)₂(2-biphenyl), resulted in a higher ratio of 66 vs. 67. The synthesis of siloles was also accompanied by the formation of the corresponding aminopinacolborane, and was extended to other silylboranes including (Me₂N)Me₂SiBpin and (pyrrolidino)Me₂SiBpin [126].

Scheme 37 Regioselective synthesis of disubstituted siloles

Moberg and co-workers subjected a number of 1,3-enynes to palladium catalyzed silaborations using PhMe₂SiBpin. The reactions required relatively high loadings of palladium and phosphine ligand, as well as stoichiometric quantities of diisobutylaluminium hydride (DIBALH). 1,2-Silaboration led to dienes 68 in all cases.
Alternatively, changing the transition metal catalyst to a platinum analogue and the 1,3-enynes substituent to a sterically hindering functionality resulted in 1,4-silaboration and isolation of the corresponding allene 69 (Scheme 38) [127].

**Scheme 38** Substrate controlled silaborations of 1,3-enynes
Most of the reports to date utilize either phosphine or isocyanide ligand sets. In their investigations into NHC-palladium catalysis, Spencer, Navarro and co-worker observed that [Pd(ITMe)_2(PhC≡CPh)] (44) catalyzed the silaborations of terminal and internal alkynes with PhMe_2SiBpin to afford a number of known and novel 1-silyl-2-boryl alkenes 54 and 70 (Scheme 39) [128]. This protocol represented the first example of alkyne silaborations employing NHC ligands. All reactions proceeded with 100% syn-stereoselectivity and, in the case of terminal alkynes 100% ‘normal’ regioselectivity. High regioselectivities were also noted when using unsymmetrical internal alkynes, with the silyl moiety favoring a position that is geminal to the aryl ring. Unprecedented mild reaction temperatures for terminal alkynes, short reactions times, and low catalyst loadings were reported.
Nickel

Ito reported the double insertion of terminal alkynes into the Si-B bond of PhMe$_2$SiBpin to afford $Z,Z$-1-silyl-4-boryl-1,3-butadiene derivatives in a regio- and stereoselective manner. The reactions proceeded using catalytic quantities of $[\text{Ni(acac)}_2]$ and the reductant DIBALH to afford a 3:1 mixture of 71 and 72 (Scheme 40). The major product 71 was a result of head-to-head dimerization of the alkyne, whereas head-to-tail dimerization gave 72. Dimerization yields were increased by using a large excess of alkyne and were retarded by the introduction of a phosphine. This protocol was also extended to internal alkynes with the exception of diphenylacetylene, which was inert under the reaction conditions. The application to diynes resulted in intramolecular cyclization and the formation of the dimethylenecyclohexane derivatives [129].

Gold

The only other metal-mediated alkyne silaboration in the literature utilized gold nanoparticles supported on titania (Au/TiO$_2$). Stratakis and co-workers used 1 mol% Au/TiO$_2$ to catalyze the silaboration of terminal alkynes at room temperature to form syn-2-boryl-1-silyl-1-alkenes 73 (Scheme 41). These alkenes were formed with opposite
or ‘abnormal’ regioselectivities with respect to the analogous palladium examples, which was attributed to the steric factors imposed by the Au nanoparticles during the 1,2-addition of the silylborane to the alkynes. Side products in this reaction were the ‘normal’ regioselective silaborated alkenes, the bis(silyl)ated alkenes and $\text{B}_2\text{pin}_2$. The presence of bis(silyl)ated alkene and $\text{B}_2\text{pin}_2$ was explained by separately stirring $\text{PhMe}_2\text{SiBpin}$ under the catalytic conditions in the absence of alkyne. The authors observed the formation of $\text{PhMe}_2\text{SiSiMe}_2\text{Ph}$ and $\text{B}_2\text{pin}_2$ as a result of metal-catalyzed silylborane metathesis, a competing reaction in this silaboration protocol. Extension to internal alkynes resulted in mixtures of regioisomers or no yield at all [130].

![Scheme 41 Gold catalyzed ‘abnormal’ silaboration of terminal alkynes](image)

**Scheme 41** Gold catalyzed ‘abnormal’ silaboration of terminal alkynes

### 2.5 Tin-Tin (Sn-Sn)

Palladium

Organostannanes are often utilized in the chemoselective formation of C-C bonds through Migita-Kosugi-Stille reactions [131]. The development of new methodologies
in the construction of C-Sn bonds is therefore of high interest. A particularly attractive
eexample is the insertion of alkynes into Sn-Sn bonds, distannation. The resulting
alkenes have two new C-Sn bonds and are frequently formed with high
stereoselectivities. Low-valent palladium complexes are regularly used to catalyse the
distannation of alkynes.

Some of the first investigations into distannation of alkynes were carried out by
Mitchell and co-workers. Hexamethyldistannane (Me₃SnSnMe₃) and terminal alkynes
were mixed in the presence of catalytic quantities of [Pd(PPh₃)₄] to form Z-1,2-distannyl
alkenes 74 (Scheme 42). Aryl, alkyl and propargyl ether substituents were tolerated.
Distannation of acetylene at elevated temperatures initially led to the Z-isomer, which
quickly isomerized to the thermally stable E-isomer. The Z to E isomerization was also
observed in the absence of catalyst under photolysing conditions [132].

![Scheme 42](image)

**Scheme 42** The first palladium catalyzed distannation of terminal alkynes

Mitchell later expanded this protocol to a wider variety of terminal alkynes including
functionalities such as alcohols, amides, esters and silyl groups. The Sn-Sn bond
precursor was also extended to other hexaalkyl distannanes (hexaethyl and hexabutyl ditin)
[133], and to 1,2,4,5-tetras-tannacyclohexanes [134]. The latter were further employed in
the distannation of trimethylstannylethyne to synthesise the first 1,1,2-trisstannylalkene

![Diagram 1](image)
derivatives 75 (Scheme 43) [135].

**Scheme 43** Distannation of trimethylstannylethyne in the synthesis of trisstanylalkenes

Piers and co-workers reported the distannation of alkyl-2-alkynylates using Me₃SnSnMe₃ and a [Pd(PPh₃)₄] catalyst in THF at room temperature (or reflux) to form Z-2,3-bis(trimethylstannyl)-2-alkenoates 76 (Scheme 44). A vast array of functionality was tolerated including alkenyls, ethers, silyl ethers and primary halides [136]. Alkenoates with an ω-halogeno-alkyl group were treated with MeLi which resulted in a transmetallation-cyclization reaction to afford 2-trimethylstannylcycloalk-1-enes 77 (Scheme 45) [137]. The distannation protocol was also extended to N,N-dimethyl-2-alkynylamides and the formation of Z-N,N-dimethyl-2,3-bis(trimethylstannyl)-2-alkenamide 76q. Compounds 76 were thermally labile and transferred upon heating or at room temperature to the thermodynamically stable E-isomers [136].

\[
R_1 \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv \equiv
Scheme 45 Transmetallation-cyclization of distannylated o-halogeno-alkenoates

The weakness of the C-Sn bond meant that it was possible to use vinyltins in electrophilic substitution reactions. Mitchell and co-workers detailed the reactivity potential of the Z-1,2-bis(trimethylstannyl)-1-alkenes with the electrophiles p-tolylsulphonylisocyanate (TSI), dichloromethylmethylether (DCME), trimethylsilyl chlorosulphonates and sulfur oxides [138].

Recently, Foucher and co-workers depicted the insertion of acetylene and phenylacetylene into the backbone of poly[di(nbutyl)]stannane. This resulted in the formation of new alkene-tin polymers 78 and 79, respectively (Scheme 46) [139].

Scheme 46 Distannation employing the backbone of poly[di(nbutyl)]stannane

Platinum

The only examples of distannation of alkynes employing a platinum catalyst were carried out by Wrackmeyer and co-workers. The distannane, 1,2-distanna-
[2]ferrocenophane reacted sequentially or in one pot with \([\text{Pt}(\text{PPh}_3)_2(\eta^2-\text{C}_2\text{H}_4)]\) and a range of terminal alkynes to form the corresponding 1,4-distanna-[4]ferroceophanes 80 (Scheme 47) [140]. Both terminal and internal alkynes were accessible. However, dimethyl acetylenedicarboxylate gave the distannation product in a side reaction while favoring cyclotrimerization to form hexamethylbenzene hexacarboxylate. Extension to analogous palladium catalysts such as \([\text{Pd}(\text{PPh}_3)_4]\) and \([\text{Pd}(\text{dba})_2]\) were unsuccessful [141].

![Scheme 47 1,2-Distanna-[2]ferrocenophane distannation of terminal alkynes](image)

**Scheme 47** 1,2-Distanna-[2]ferrocenophane distannation of terminal alkynes

**Copper**

In 2013, Yoshida carried out the first catalytic distannation of alkynes using a copper catalyst. \([\text{Cu(OAc})(\text{PPh}_3)_3]\) in the presence of \(\text{Cs}_2\text{CO}_3\) was used to optimize the reaction between \(\text{Me}_3\text{SnSnMe}_3\) and 1-octyne affording 81. The authors then managed to distannylate 1-hexyne, 1-decyne and branched aliphatic terminal alkynes bearing isoamyl, isobutyl and cyclopentyl, as well as chloro, amino and cyano functionalities. Alkynes that were sterically congested resulted in sluggish reactions and low yields. It was proposed that the reaction proceeded through a Cu-Sn bonded intermediate 15 derived from a CuOR’ complex and a base-activated distannane. Subsequent addition of 15 to a C-C triple bond afford \(\beta\)-stannylalkenyl copper species 16, which is then
recaptured with Me$_3$SnOR’ to give the 1,2-distannylated alkene with regeneration of the CuOR’ complex (Scheme 48) [142].

**Scheme 48** Copper catalyzed distannation of 1-octyne and proposed mechanism

**2.6 Tin-Silicon (Sn-Si)**

Palladium

Sn-Si bond (silylstannation) addition to alkynes results in the formation of alkenes with a new C-Sn and C-Si bond, often in a regio- and stereoselective manner. Palladium catalyzed silylstannation of alkynes are by far the most reported examples within the literature and have found application in the synthesis of natural products [143,144], and pharmaceuticals [145]. The first palladium catalyzed examples of alkyne silylstannations were shown by Mitchell and co-workers. The authors reacted a range of
terminal alkynes with \((\text{trimethylsilyl})\text{trimethylstannane} (\text{Me}_3\text{SiSnMe}_3)\) in the presence of \([\text{Pd(PPh}_3]_4\) under solvent-free conditions to yield the corresponding \(Z\)-1-silyl-2-stannyl-1-alkenes \((82)\) (Scheme 49). In all cases the silyl moiety added regioselectively at the terminal carbon \([146]\).

\[
\begin{align*}
\text{R} & \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \quad \ quad
Scheme 50 Palladium(0)-catalyzed silylstannation followed by regioselective cyclization

Ito later accomplished the silylstannation of 1-alkoxyalkynes employing the precatalytic combination of [Pd(OAc)$_2$]/tert-octylisocyanide. The reactions proceeded at room temperature and yielded the syn-addition products 86, with the silyl group regioselectively introduced at the carbon atom bearing the alkoxy moiety (Scheme 51).

\[
\text{Scheme 51 Silylstannation of 1-alkoxyalkynes}
\]

[Pd(PPh$_3$)$_4$] was inactive in these transformations at both room and elevated temperatures [148]. The resulting alkene adducts were then exposed to a range of reactions including Stille cross-couplings, iodination at the C-Sn bond and the formation of acylsilanes.

Singer reported the silylstannation of terminal alkynes with Bu$_3$SnSiMe$_2$Ph using catalytic quantities of [Pd(PPh$_3$)$_4$] immobilised in the ionic liquid 1-"butyl-3-
methylimidazolium hexafluorophosphate ([bmim][PF$_6$]). High stereo- and regioselectivities were observed and simple ether extraction resulted in isolation of the Z-1-silyl-2-stannyl-1-alkenes 87 (Scheme 52) [149]. The palladium(0)-ionic liquid combination was recyclable with no loss of activity even after 10 cycles [150].

The bismetallative cyclization of 1,3-enynes, using Bu$_3$SnSiMe$_3$, is dependent on the choice of ligand and palladium source. The use of [Pd(PPh$_3$)$_4$] results in ‘normal’ silylstannation of the alkyne affording 88. However, upon removing the phosphine and using [Pd$_2$(dba)$_3$.CHCl$_3$] or [Pd(OH)$_2$/C], cyclized compounds 89 can be isolated as the major products of the reaction (Scheme 53). The bismetallative cyclization can also be observed on employing nucleophilic N-heterocyclic carbenes with bulky alkyl N-substituents [151].

Scheme 52 Alkyne silylstannation using ionic liquid immobilised palladium(0)
It is possible to tune the regioselectivity of alkyne silylstannation by changing the ligand of a palladium(II) catalyst. Treatment of fluorine containing internal alkynes with Bu$_3$SnSiMe$_3$ in the presence of 2.5 mol% [Cl$_2$Pd(PPh$_3$)$_2$] yields the silylstannylated adducts 90. However, by switching the palladium catalyst to [Cl$_2$Pd(tBuNC)$_2$], the opposite regioselectivities 91 are observed (Scheme 54) [152].

Scheme 53 Silylstannation and bis metallative cyclization of 1,3-enynes
Scheme 54 Palladium catalyst-dependent regioselective silylstannation

Copper
The only example of a copper-catalyzed alkyne silylstannation was carried out by Yoshida and co-workers. The authors detailed a three-component coupling reaction employing terminal alkynes, a silylborane (PhMe₂SiBpin) and a tin alkoxide (⁶Bu₃SnO⁵Bu) in the presence of a Cu(I) catalyst ([CuCl-P₃Bu₃]). The observed regioselectivities were inverse to those of conventional silylstannation under palladium catalyzed conditions, with the stannyl moiety predominantly adding to the terminal carbon as shown in 92. A range of alkyl branched and unbranched alkynes bearing cyano, bromo, hydroxyl or amino functionalities were accessible (Scheme 55). The authors proposed a similar mechanism to the distannation of alkynes mediated by a copper(I) catalyst. A silylcopper species, CuSiMe₂Ph, is initially formed via a sigma-bond metathesis between a copper alkoxide and a silylborane. An alkyne would then insert into the Cu-Si bond to give a β-silylalkenylcopper intermediate, which is subsequently trapped by a tin alkoxide to furnish the silylstannation alkene adduct and regenerated the copper alkoxide [153].
Scheme 55 Cu(I) catalyzed ‘abnormal’ silylstannation of terminal alkynes

2.7 Tin-Boron (Sn-B)

Palladium

The borylstannation of alkynes results in the formation of alkenes with a new C-Sn and C-B bond. The first palladium-catalyzed example was shown by Tanaka and co-workers in 1996. The borylstannane 1,3-dimethyl-2-(trimethylstannyl)-2-bora-1,3-diazaacyclopentane (Me₃SnB[NMe(CH₂CH₂)NMe]) was added to alkynes using catalytic quantities of [Pd(PPh₃)₄] (Scheme 56). The reagents were added together in benzene at 0 °C and then warmed to room temperature. Terminal alkynes yielded syn-1-boryl-2-stannyl-1-alkenes 93 as the sole product. Internal alkynes were also accessible, although a higher temperature (80 °C) was necessary [154]. Weber later extended this protocol to the more sterically hindered borylstannane, 1,3-di-tert-butyl-2-(trimethylstannyl)-2-bora-1,3-diazaacyclopentane (Me₃SnB[N^tBu(CH₂CH₂)N^tBu]) [155].

Scheme 56 Palladium(0)-catalyzed borylstannation of alkynes
RajanBabu developed Tanaka’s methodology and extended it to the borylstannation of 1,3-enynes. This protocol resulted in the isolation of the syn-1-boryl-2-stanny-1-alkenes (94) in chemo-, regio- and stereoselective fashions with no complications arising due to the adjacent alkene. However, the boryl group in 94 was hydrolytically unstable. In situ treatment with pinacol and p-toluenesulfonic acid (PTSA) yielded the hydrolytically stable 95 (Scheme 57) [156].

**Scheme 57** Borylstannation of 1,3-enynes followed by boryl alcoholysis

Copper

Yoshida detailed the copper(II) catalyzed borylstannation of alkynes. A three-component coupling reaction employed an alkyne, a diboron (B$_2$pin$_2$) and a tin alkoxide ("BuSnOMe) with the aid of a copper(II)acetate/tricyclocylhexylphosphine combination. A range of internal and terminal alkynes were accessible. Internal alkynes with one aryl and one alkyl substituent resulted in perfect regioselectivities with the boryl moiety geminal to the alkyl group. In the case of terminal alkynes, the boryl group added to the terminal carbon [157]. All reactions proceeded at room temperature with catalyst loadings as low as 1 mol%. The authors proposed that these reactions proceeded through a similar mechanism to that of the silylstannation [153].
2.8 Sulfur-Sulfur and Selenium-Selenium (S-S and Se-Se)

Palladium

Organochalcogens are known to exhibit a range of pharmacological activity profiles including as potential anticancer [158], anti-inflammatory [159], and antibacterial agents [160]. The introduction of chalcogens into alkynes to form 1,2-bis(chalcogen)alkenes is challenging. The formation of 1,2-bis(chalcogen)alkenes often requires the use of heavy metals, high temperatures and results in a mixture of stereoisomers. Such methods include the reaction vinyldichlorides with thiolate anions [161] and radical reactivity between chalcogen species and alkynes [162]. The transition-metal catalyzed addition of dichalcogens to alkynes is a possible alternative to synthesizing 1,2-bis(chalcogen)alkenes in a stereoselective and atom-economical manner.
Sonoda reported the first palladium mediated addition of diaryl disulphides and diselenides to terminal alkynes to yield the corresponding Z-1,2-bis(arylthio) and Z-1,2-bis(arylseleno)-1-alkenes (96), respectively (Scheme 58). This protocol tolerated functionalities such as hydroxyl, trimethylsilyl and amino groups. The inclusion of a carbon monoxide (CO) atmosphere in these reactions lead instead to the isolation of the carbonylative addition adducts Z-1,3-bis(arylchalcogen)-2-alken-1-ones (97) (Scheme 58). A stepwise attempt at carbonylation of 96 with CO to yield 97 resulted in the isolation of only 96, suggesting that CO insertion was part of dichalcogen additions in the first instance [163].
Gareau and co-workers established a procedure that effectively introduced ‘dialkyl’-disulphides in the dithiolation of alkynes. Upon protecting the disulphide with a bulky silyl group, the dithiolation of terminal alkynes with bis(triisopropylsilyl)disulphide resulted in the isolation of the corresponding Z-1,2-bis(thio)alkenes (98). Subsequent treatment with tetra-nbutylammonium fluoride (TBAF) in the presence of excess methyl iodide (MeI) de-protected/alkylated the sulfur atoms affording 99 (Scheme 59) [164]. Additionally, Gareau investigated the reactivity of 98 towards other electrophiles including halides, epoxides and acyl chlorides [165]. Furthermore, treatment of 98 with HCl in the presence of a Lewis acid ([Zn{OTf}_2]) yielded the bicyclic adducts 2,5,7-trithiabicyclo[2.2.1]heptane (100, Scheme 60) [166].

Beletskaya and co-workers reported an alternative methodology in the addition of S-S and Se-Se bonds to alkynes. Diphenyl disulphide (Ph2S2) and diphenyl diselenide

\[
\text{R} = \begin{array}{c}
\text{Si(Pr)}_3 \\
\text{S-S} \\
\text{Si(Pr)}_3
\end{array} \quad 5 \text{ mol\% } [\text{Pd}(\text{PPh}_3)_4] \quad \text{benzene, reflux} \quad \text{(Si(Pr)}_3\text{SiS(Si(Pr)}_3\text{)}\text{)}
\]

\[98/99: R = \begin{array}{c}
\text{MeS} \\
\text{CH}_2\text{Ph} \\
\text{cyclohexyl} \\
f\text{Bu}
\end{array}\]

\[98f/99f: R = \begin{array}{c}
\text{C(CH}_3)_2\text{OH} \\
\text{CH}_2\text{Ph} \\
\text{cyclohex-1-enyl} \\
\text{C(CH}_3)_2\text{CO}_2\text{Ph}
\end{array}\]

\[98g/99g: R = \begin{array}{c}
\text{C(CH}_3)_2\text{Cl} \\
\text{C(CH}_3)_2\text{Ph}
\end{array}\]

\[98h/99h: R = \begin{array}{c}
\text{C(CH}_3)_2\text{Cl} \\
\text{C(CH}_3)_2\text{Ph}
\end{array}\]

\[98i/99i: R = \begin{array}{c}
\text{Ph} \\
\text{C(CH}_3)_2\text{CH}_2\text{Ph}
\end{array}\]

\[98j/99j: R = \begin{array}{c}
\text{C(CH}_3)_2\text{Cl} \\
\text{C(CH}_3)_2\text{Ph}
\end{array}\]

**Scheme 58** Dichalcogen and carbonylative dichalcogen additions to alkynes

**Scheme 59** Synthesis of Z-1,2-bis(methylthio)alkenes via silyl-protected dichalcogens
Scheme 60 Reactivity of 98 with acid/Lewis acid combination forming 2,5,7-trithiabicyclo[2.2.1]heptane adducts

(Ph₂Se₂) were reacted with a variety of terminal alkynes in the presence of catalytic quantities of [Cl₂Pd (PPh₃)₂], PhEH (E = S or Se) and triethylamine (NEt₃) to yield the Z-1,2-bis(arylthio) and Z-1,2-bis(arylseleno)-1-alkenes. Both PhEH and NEt₃ were essential for the success of the reaction. The yields increased on the addition of excess PPh₃, which contradicts the general trend observed for other E-E additions to alkynes within the literature [167]. It was later shown that excess PPh₃ prevented the rapid polymerization to [Pd(EAr)₂]ₙ and therefore the inhibition of the palladium catalyst [168].

Beletskaya also reported the dithiolation of terminal alkynes utilizing a Pd(0) catalyst supported by a triphenylphosphine resin under conventional [169], and microwave heating conditions [170]. Simple filtration resulted in the isolation of the Z-1,2-bis(thio)-1-alkenes. This approach was not applicable to diaryl diselenides. Other palladium(0) supported mediators for the dithiolation of terminal alkynes include the MCM-41-supported bidentate phosphine Pd(0) catalyst reported by Cai [171].
Zeni described the addition of diorganyl diselenides and disulfides to terminal alkynes in the presence of an iron(III) chloride (FeCl$_3$) catalyst. The best results were observed using diaryl diselenides bearing neutral electron-donating and withdrawing groups. The electronic nature of the terminal alkyne substituent did not have an effect on the rate or yield of the reaction [172]. Iron catalyzed addition to 1,4-butyn-diols, pentyne-1,5-diol and 4-amino butynol afforded 3,4-bis(organochalcogen)-2,5-dihydrofurans (101), 4,5-bis(organochalcogen)-3,6-dihydro-2H-pyrans (102) and 2,5-dihydro 1H-pyrrole derivatives (103), respectively, under mild aerobic conditions (Scheme 61) [173]. 1,3-Diynes in the presence of dibutyl diselenide or dimethyl disulfide and stoichiometric quantities of FeCl$_3$ yielded symmetrical and unsymmetrical 3,4-bis(butylselanyl)chalcogenophanes (104). In the synthesis of the 104, the cyclization was stereoselective providing exclusively the desired E-selenoenynes as intermediates. The selenophanes then formed via an intramolecular 5-endo-dig cyclization [174].

Copper

The addition of the catalytic mixture of CuI, zinc dust and glycerol resulted in the stereoselective addition of diaryl dichalcogenides to form a variety of E-1,2-bis-chalcogen alkenes (105) (Scheme 62). Zinc and glycerol were essential to the reaction; Zn reduced Cu(I) to Cu(0) while glycerol acted as a solvent, but also as a possible reducing agent for the reduction of Zn(II) to Zn(0) [175].
Scheme 61 Synthesis of bis(organochalcogen)-dihydrofurans, dihydro-2H-pyrans, dihydro 1H-pyrroles and selenophanes
Scheme 62 Copper catalyzed E-dithiolation and diselenation of terminal alkynes

Nickel

The only examples of nickel-catalyzed addition of diaryldisulphides to alkynes were developed by Beletskaya and co-workers. The use of 3 mol% [Ni(acac)$_2$] and 30 mol% PMePh$_2$ at 100 °C under solvent-free conditions resulted in the stereoselective dithiolation of both internal and terminal alkynes to form Z-dithiolated alkene products. The reaction temperature was important: too low meant incomplete reactivity and too high led to a mixture of stereoisomers [176].

Rhenium

The stoichiometric reaction between the tetrathiometallate anion [ReS$_4$]$^-$ and diphenylacetylene, 2-butyne and bis(trimethylsilyl)acetylene in the presence of elemental sulfur yielded the dithiolation adducts 106 (Scheme 63) [177].

Scheme 63 Rhenium mediated stoichiometric dithiolation of internal alkynes

Rhodium
Yamaguchi and co-workers showed that it was possible to dithiolate terminal alkynes using the dialkyl disulphide, Bu$_2$S$_2$, employing catalytic quantities of a Rh-phosphine complex, tris(p-methoxyphenyl)phosphine and trifluoromethane sulfonic acid affording the corresponding Z-bis(alkythio)alkenes (107) (Scheme 64). A range of functionality at the terminal alkyne substituent was accessible including hydroxyl, tert-butylidimethylsiloxy and nitrile. However, internal alkynes were not accessible with this protocol [178].

![Scheme 64 Rhodium(I) catalyzed dithiolation of terminal alkynes](image)

Yamaguchi and co-workers extended their studies to the addition of disulphides and diselenides to alkynes in cross-over experiments. A 1:1 mixture of diaryl disulphides and diaryl diselenides were reacted with terminal alkynes using the same Rh complex and 1,1'-bis(diphenylphosphino)ferrocone (dpff). This resulted in the formation of Z-1-arylseleno-2-(arylthio)-1-alkenes (108) as the major product (Scheme 65). The amounts of minor by-products Z-2-arylseleno-1-(arylthio)-1-alkene, Z-1,2-bis(arylthio)alkene and Z-1,2-bis(arylseleno)alkene were insubstantial. However, the minor product ratio became significant upon removal of trifluorosulfonic acid or when increasing the steric hindrance surrounding the alkynes [179].
Nakamura and co-workers developed the AuCl-catalyzed cyclization of (o-alkynylphenylthio)silanes (109) to form the corresponding 3-silylbenzo[b]thiophenes (110). The reaction was proposed to proceed initially by coordination of the gold species to the alkynyl moiety. The sulfur atom then acts as an intramolecular nucleophile, attacking the electron deficient alkyne which results in a silylsulfonium intermediate. Subsequently, [1,3]-migration of the silyl group and elimination of AuCl yielded 110 (Scheme 66). The yield was highly dependent on the natural of the alkyne substituents with electron rich aromatic rings producing higher yields than electron poor or bulky groups (which inhibited the reaction) [180].

2.10 Sulfur-Boron (S-B)

Palladium

Suzuki and Miyuara employed 9-(alkylthio)-9-borabicyclo[3.3.1]nonane in the palladium(0) catalyzed thioboration of terminal alkynes to produce 9-[Z-2-(alkylthio)-1-alkenyl]-9-borabicyclo[3.3.1]nonane derivatives. These reactions were highly regio-
Scheme 66 Gold(I) catalyzed cyclization of (o-alkynylphenylthio)silanes

and stereoselective with the boryl group adding to the terminal carbon in all cases. The reactions were sufficiently mild that a variety of functionalities were tolerated [181].

2.11 Germanium-Germanium (Ge-Ge)

Palladium

In contrast to Si-Si and Sn-Sn bonds, the insertion of alkynes into Ge-Ge bonds has been investigated to a much lesser extent. The resulting compounds are expected to have a reactivity profile somewhere in-between their Si-Si and Sn-Sn analogues. The majority of alkyne digermylations in the literature are palladium-catalyzed. The first example was reported by Ando and co-workers. In their work, a strained cyclic digermirane was reacted with acetylene and dimethyl acetylenedicarboxylate in the presence of 10 mol% [Pd(PPh₃)₄] resulting in the formation of the digermacyclopentene
111. When X was a sulfur atom, it was possible to selectively cleave the Ge-S bond to afford 112 (Scheme 67) [182].

Scheme 67 Linker-atom dependent addition strained digermane to alkynes

Mochida and co-workers reacted 1,1,2,2,3,3,4,4-octaisopropyltetragermetane (\(\text{Me}_2\text{Ge}\)) with various terminal alkynes in the presence of palladium complexes to synthesise 1,2,3,4-tetrahydro-1,2,3,4-tetragermins (113), \(\Delta^4\)-1,2,3-trigemolene (114) and 1H-germoles (2,4-, 3,4- and 2,3-disubstituted) (115) (Scheme 68). The yields of 114 and 115 increased with time, a fact attributed to the thermolyis of 113 in the presence of excess alkyne. The formation of 114 from 113 suggested extrusion of diisopropylgermylene (iPrGe:), which was readily trapped by two equivalents of alkyne to give 115 [183].

In 1991, Tanaka used linear non-strained digermanes in the digermylation of alkynes. 1,2-Dichloro-1,1,2,2-tetramethyldigermane was reacted with phenylacetylene in the presence of a palladium(0) catalyst to form 1.111 (Scheme 1.66). The extension of
the protocol to hexamethyldigermane resulted in very low conversions [184]. Following studies accomplished the conversion of 116 to 1,2-digermacyclobut-2-enes (117) by reductive cyclization in the presence of sodium metal. The treatment of 117 with alkynes in the presence of palladium catalysts resulted in the digermylation and the formation of the corresponding 1,4-digermacyclohex-2,5-dienes 118 and 119 (Scheme 69) [185].

**Scheme 68** Addition of a tetragermetane to internal alkynes
Scheme 69 Digermylation, reductive cyclization and digermylation employing
dichlorodigermanes and terminal alkynes

Platinum

The digermylation of terminal with hexamethyldigermane has only been accessible
employing a platinum catalyst at $120 \degree C$, affording the corresponding $Z$-1,2-
bis(trimethylgermyl)ethenes (120) (Scheme 70). Lowering the temperature resulted in
deterioration of the yields [186]. Internal alkynes were unreactive.

\[ \begin{align*}
R_1 & \equiv R_2 + \text{Me}_3\text{Ge} - \text{GeMe}_3 \\
& \xrightarrow{5 \text{ mol\% } \text{Pt(acac)}_2} \text{toluene, } 120 \degree C \\
120a: R_1 = \text{Ph}, R_2 = \text{H} & \quad 120c: R_1 = \text{SiMe}_3, R_2 = \text{H} \\
120b: R_1 = \text{\textsuperscript{t}Bu}, R_2 = \text{H} & \quad 120d: R_1 = \text{CO}_2\text{Me}, R_2 = \text{H}
\end{align*} \]

Scheme 70 Platinum catalyzed digermylation of alkynes with hexamethyldigermane

2.12 Germanium-Tin (Ge-Sn)

Palladium

Piers and co-workers reported the germylstannation of $\alpha,\beta$-acetylenic esters with
Bu$_3$SnGeMe$_3$ to afford E-2-(tri-$\text{\textsuperscript{t}Bu}$stannyl)-3-(trimethylgermyl)alk-2-enoates (121)
as the major product. The reactivity of the resulting germyl and stannyl groups were
separately assessed. 121 was treated with BuLi and an alkyl halide to form 122 via the
transmetallation of Bu$_3$Sn. The germyl moity was also transformed into a C-I bond
upon addition of iodine (Scheme 71) [187].

Nakano synthesized $Z$-1-aryl-2-germyl-1-stannylethenes (123) by adding
tributyl(triethylgermyl)stannane to aryl terminal alkynes in the presence of catalytic
amounts of [Pd(dba)$_2$] and 4-ethyl-1-phospha-2,6,7-trioxabicyclo[2.2.2]octane (Scheme
This protocol was extended to ethynylthiophene and 2-methyl-3-butyn-2-ol. The regioselective addition of the germyl moiety at the terminal carbon was favored in all cases [188].

Scheme 71  E-Germylstannation of α,β-acetyleneic esters

Scheme 72  Z-Germylstannation of aryl terminal alkynes

2.13 Germanium-Boron (Ge-B)

Nickel, Palladium and Platinum

In their investigation into the silaborative dimerization of alkynes catalyzed by nickel complexes, Ito and co-workers developed the analogous germylborane reaction. The
product/s obtained in the germylboration of 1-hexyne were highly dependent on the metal catalyst used. In the presence of [Ni(acac)$_2$/DIBALH, the germylbated dimerized product 124 was obtained. By altering the catalyst to [Pd(OAc)$_2$/isocyanide, a 1:1 mixture of 124 and the germylboration adduct 125 was isolated, whereas catalytic-quantities of Pt(PPh$_3$)$_2$(C$_2$H$_4$) resulted in exclusively 125 (Scheme 73) [129].

Scheme 73 Catalyst dependent germylboration of 1-hexyne

3 Conclusions

This review presents the state-of-the-art in homo- and heterogeneous transition metal catalyzed hetero element-element’ additions to alkynes. These reactions yield highly functionalized multi-substituted alkenes with high regio- and stereoselectivities. The early literature was limited to reactive or unstable E-E’ bonds, and reactions were primarily mediated by platinum group transition metal complexes with phosphine or isocyanide ligands. Developments over the past decade made it possible to employ other
transition metals such as coinage metal complexes, and utilize commercially available
E-E’ bonds that are air and moisture stable. Furthermore, the application of NHCs as
ligands has vastly improved conditions in an unprecedented manner. However, E-E’
bonds are still restricted in their application and the state-of-the-art catalysts remain
flawed. Continuing efforts are likely to focus on earth abundant transition metal
complexes and new E-E’ bonds that improve current regio- and stereoselectivities.

4 References
137 (1986) 513
Ethier, J.F. Evans, A.W. Ford-Hutchinson, J.Y. Gauthier, R. Gordon, J. Guay, M.
Gresser, S. Kargman, B. Kennedy, Y. Leblanc, S. Leger, J. Mancini, G.P.
O’Neill, M. Ouellet, M.D. Percival, H. Perrier, D. Riendeau, I. Rodger, P. Tagari,
M. Therien, P. Vickers, E. Wong, I.-J. Xu, R.N. Young, R. Zamboni, Bioorg.
111 (1989) 643
[8] H. Wiseman, Tamoxifen: Molecular Basis of Use in Cancer Treatment and
Prevention, Wiley, Chichester, 1994
[29] D. Canseco-González, V. Gómez-Benítez, O. Baldovino-Pantaleón, S.


(2007) 5929

Ito, Organometallics 27 (2008) 1736


[38] H. Sakurai, K. Mochida, M. Kira, J. Am. Chem. Soc. 97 (1975) 931

Chem. 199 (1980) 43

(1981) 149


Organometallics 20 (2001) 5537
(1993) 3933
32 (1993) 1473
[54] H.W. Peindy N’dongo, S. Clément, S. Richeter, F. Guyon, M. Knorr, P. Le
Int. Ed. 54 (2015) 5579
645 (2002) 47
1597


Soc. 124 (2002) 1584


Biomol. Chem. 11 (2013) 1199
[176] V.P. Ananikov, K.A. Gayduk, N. V. Orlov, I.P. Beletskaya, V.N. Khrustalev,
7219
(1991) 245
(2005) 2967
